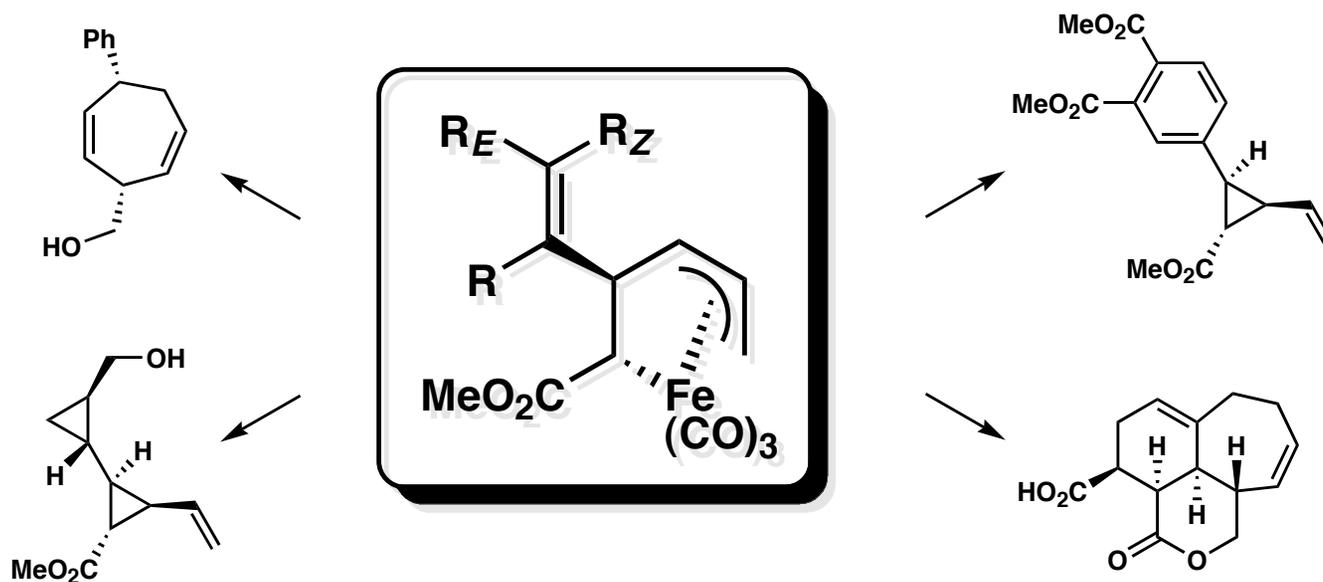


Reactivity of (2-Alkenyl-3-pentene-1,5-diyl)iron Complexes: Preparation of Functionalized Vinylcyclopropanes and Cycloheptadienes

Rajesh K. Pandey, Lizhu Wang, Nathaniel J. Wallock, Sergey Lindeman, and William A. Donaldson*

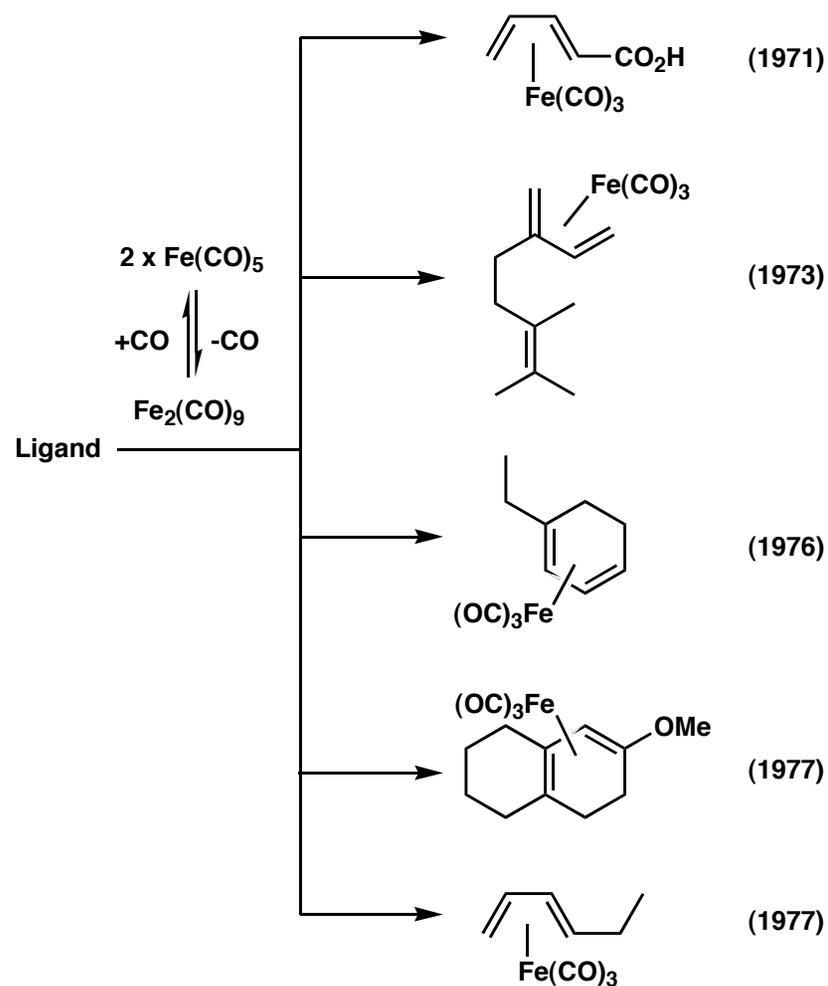
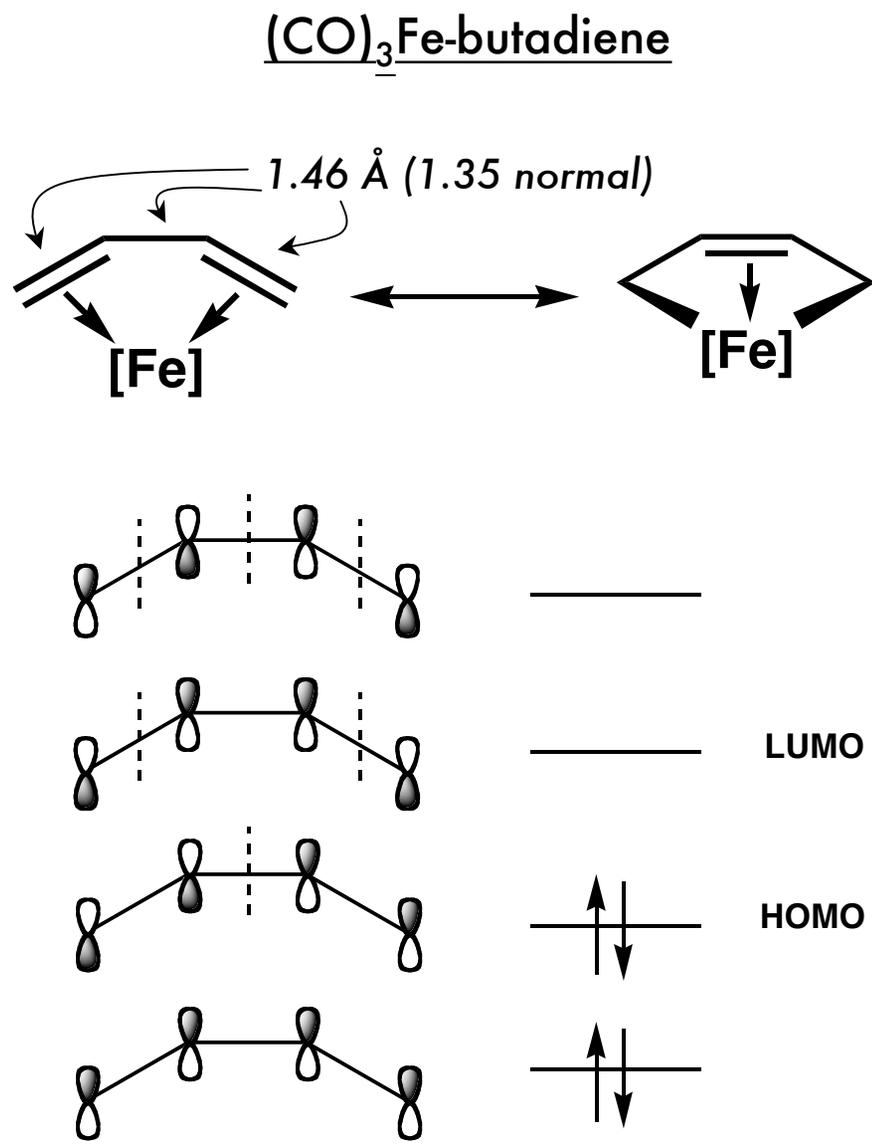
J. Org. Chem., 2008, 73 (18), 7236–7245



Adam T. Hoye
Current Literature
Sept. 27th, 2008

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Dienyliron complexes

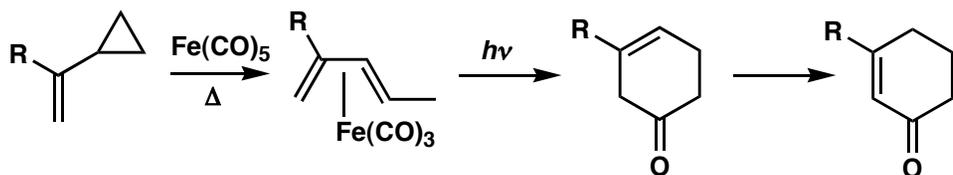


Crabtree, R. H. *The Organometallic Chemistry of the Transition Metals*. 3rd ed. Wiley & Sons, NY. 2001.

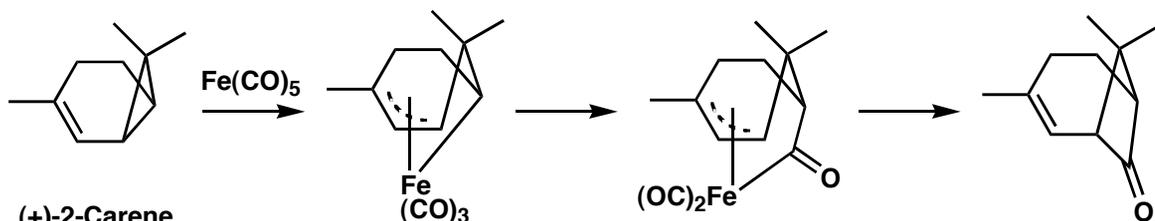
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Uses of dienyliron complexes

Rearrangements and CO-insertions



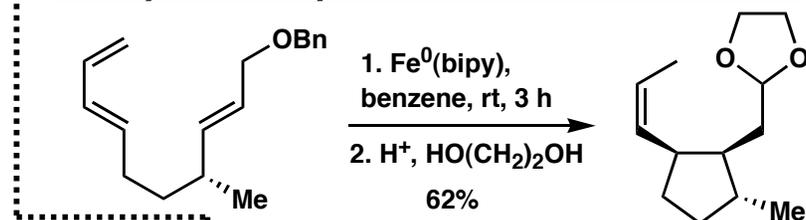
Aumann, R. *J. Am. Chem. Soc.* **1974**, *96*, 2631



(+)-2-Carene

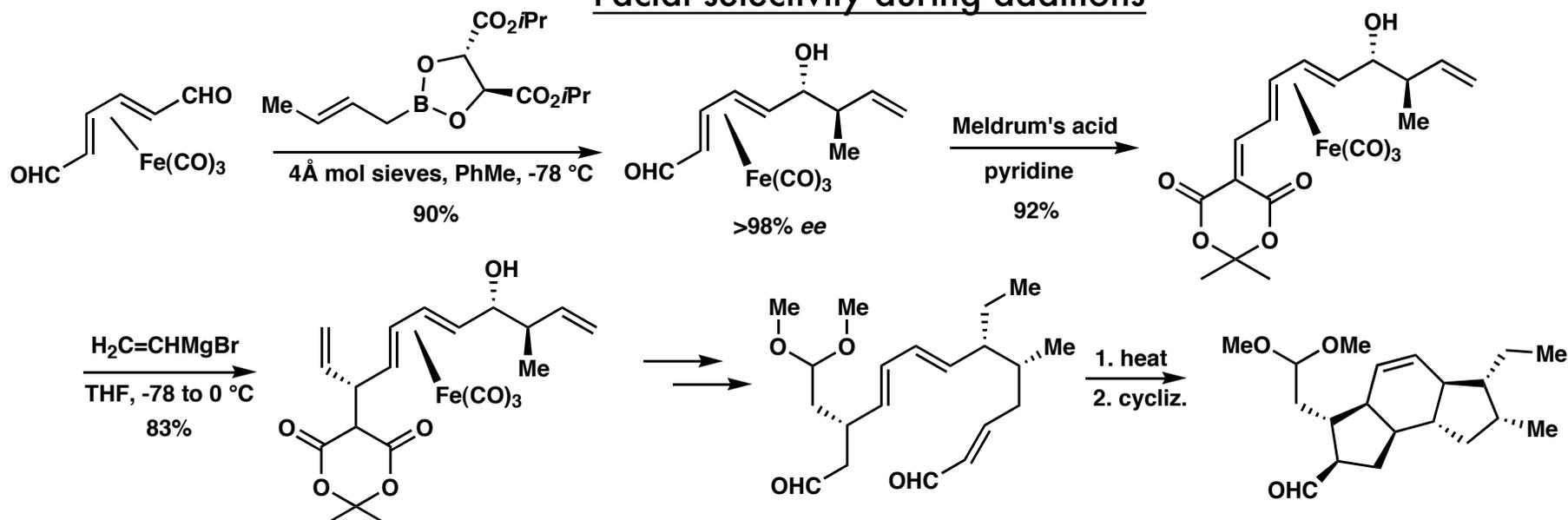
Eilbracht, P.; Winkels, I., *Chem. Ber.*, **1991**, 191.

Pericyclic catalysis (cisoid conformation)



Takacs, J. M.; Anderson, L. G.;
Creswell, M. W. Takacs, B. E.
Tetrahedron Lett. **1987**, *28*, 5627

Facial selectivity during additions

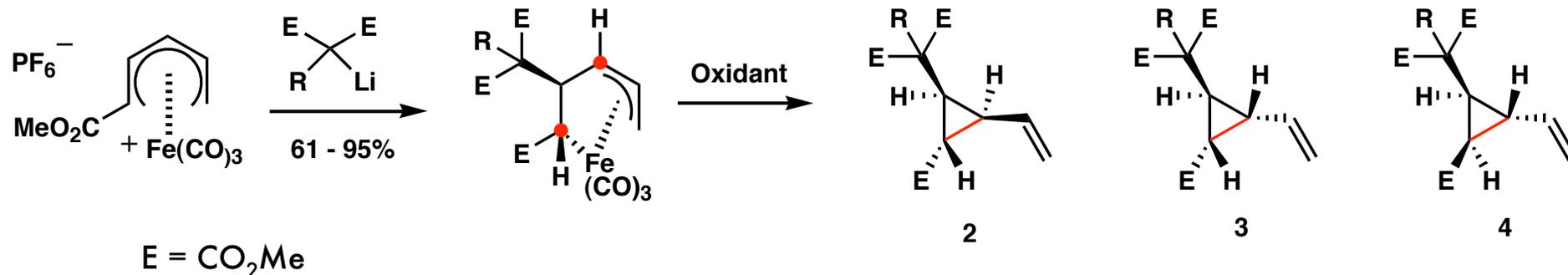


Roush, W. R.; Wada, C. K., *J. Am. Chem. Soc.* **1994**, *116*, 2151-2152

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Donaldson group previous work

Vinylcyclopropanes from reductive elimination:

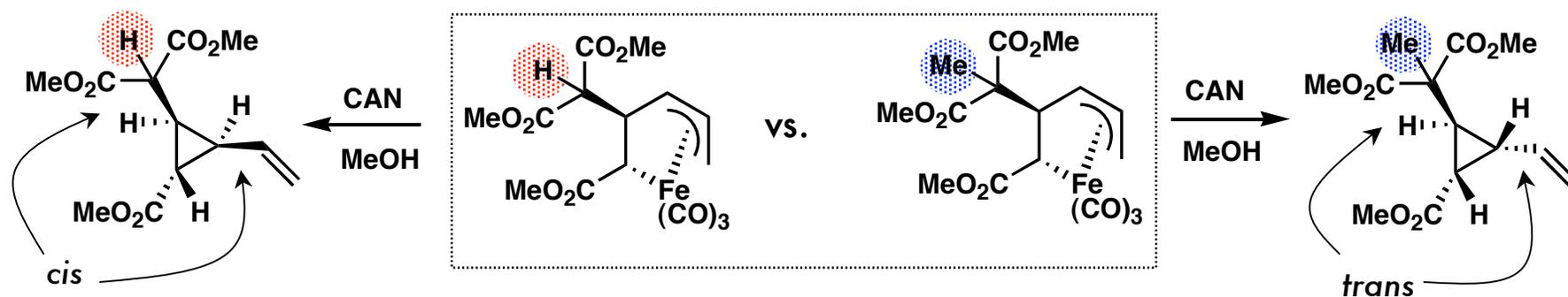


entry	R	conditions	product ratio (2:3:4)	total yield (%)
1	H (a)	CAN/DMF/23 °C	10:1:0	70
2	H (a)	TMANO/C ₆ H ₆ /80 °C	2:4:1	69
3	Me (b)	CAN/DMF/23 °C	0:1:0	55
4	Me (b)	TMAO/C ₆ H ₆ /80 °C	0:1:0	56
5	OMe (c)	CAN/DMF/23 °C	0:1:0	25

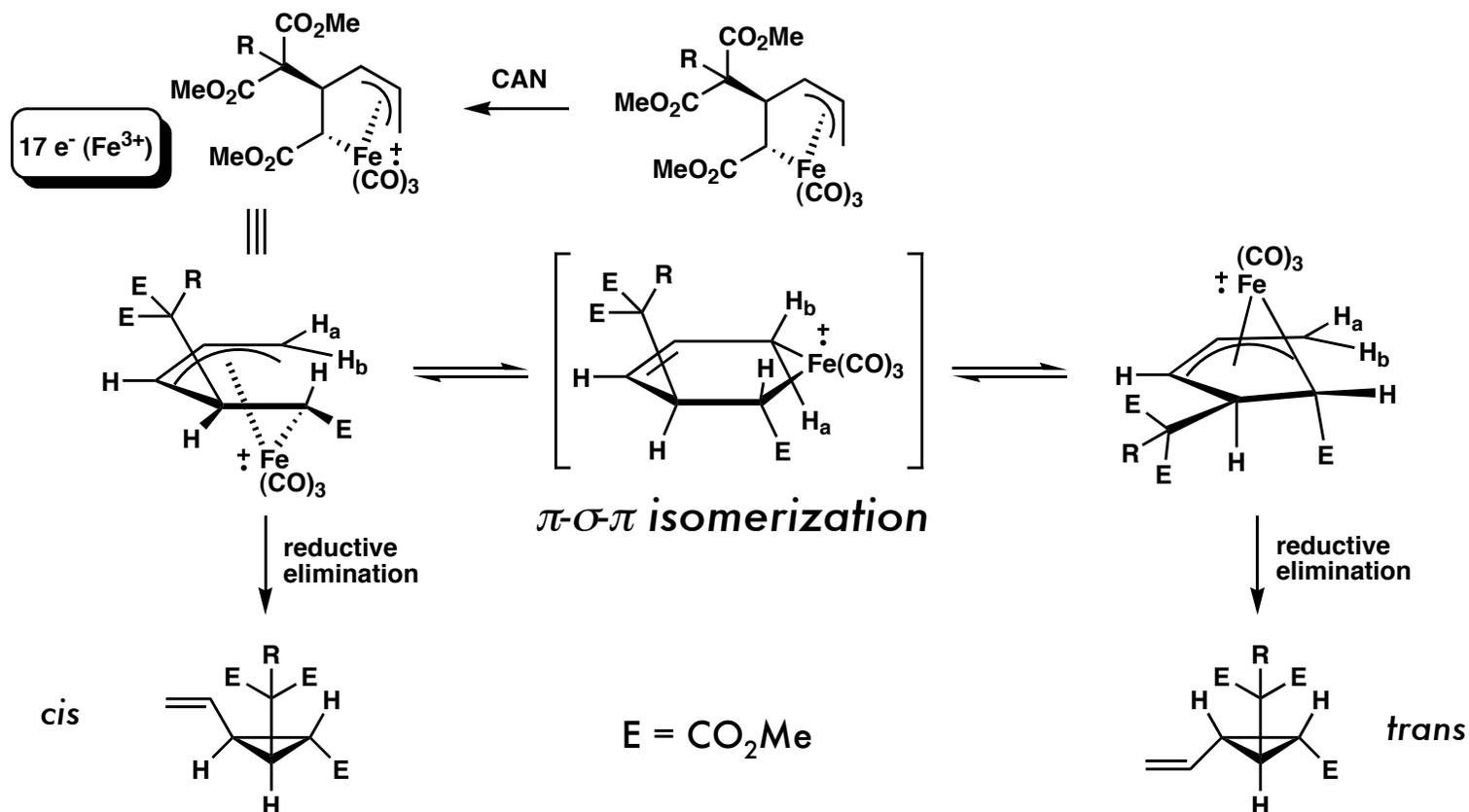
TMANO = Me₃NO

Different diastereoselectivities based on malonate substituent **AND** temperature

Diastereoselectivity in reductive elimination



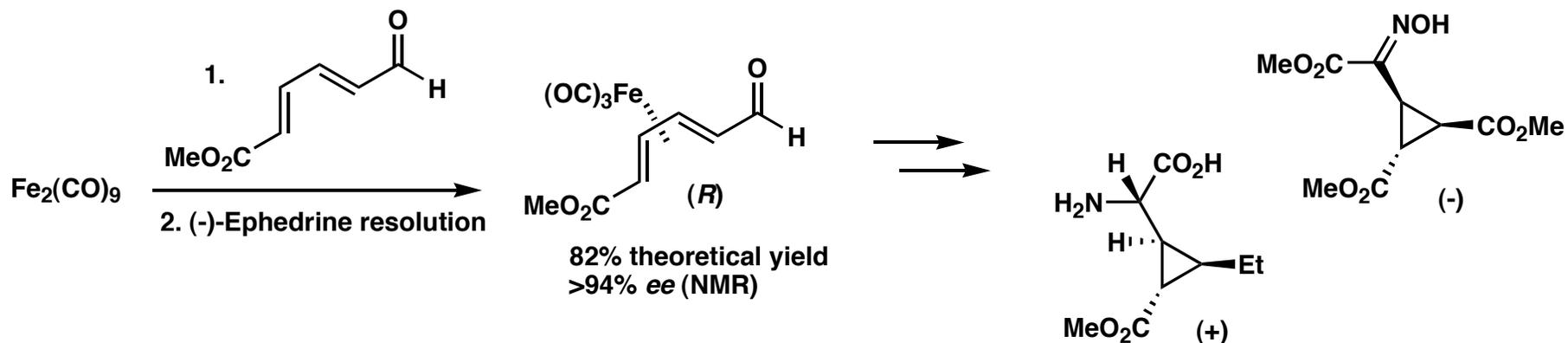
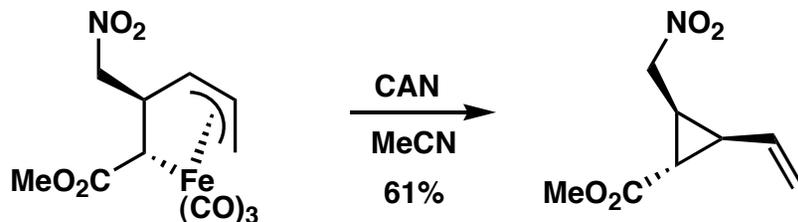
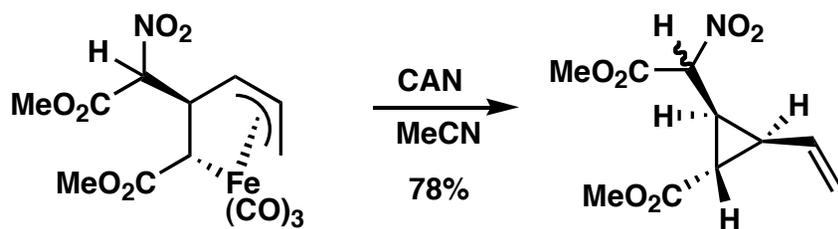
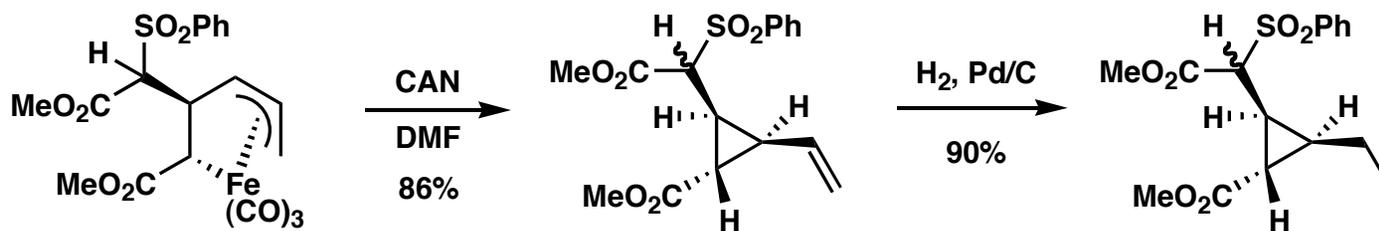
Proposed Mechanism:



Yun, Y. K.; Donaldson, W. A. *J. Am. Chem. Soc.* **1997**, *119*, 4084-4085

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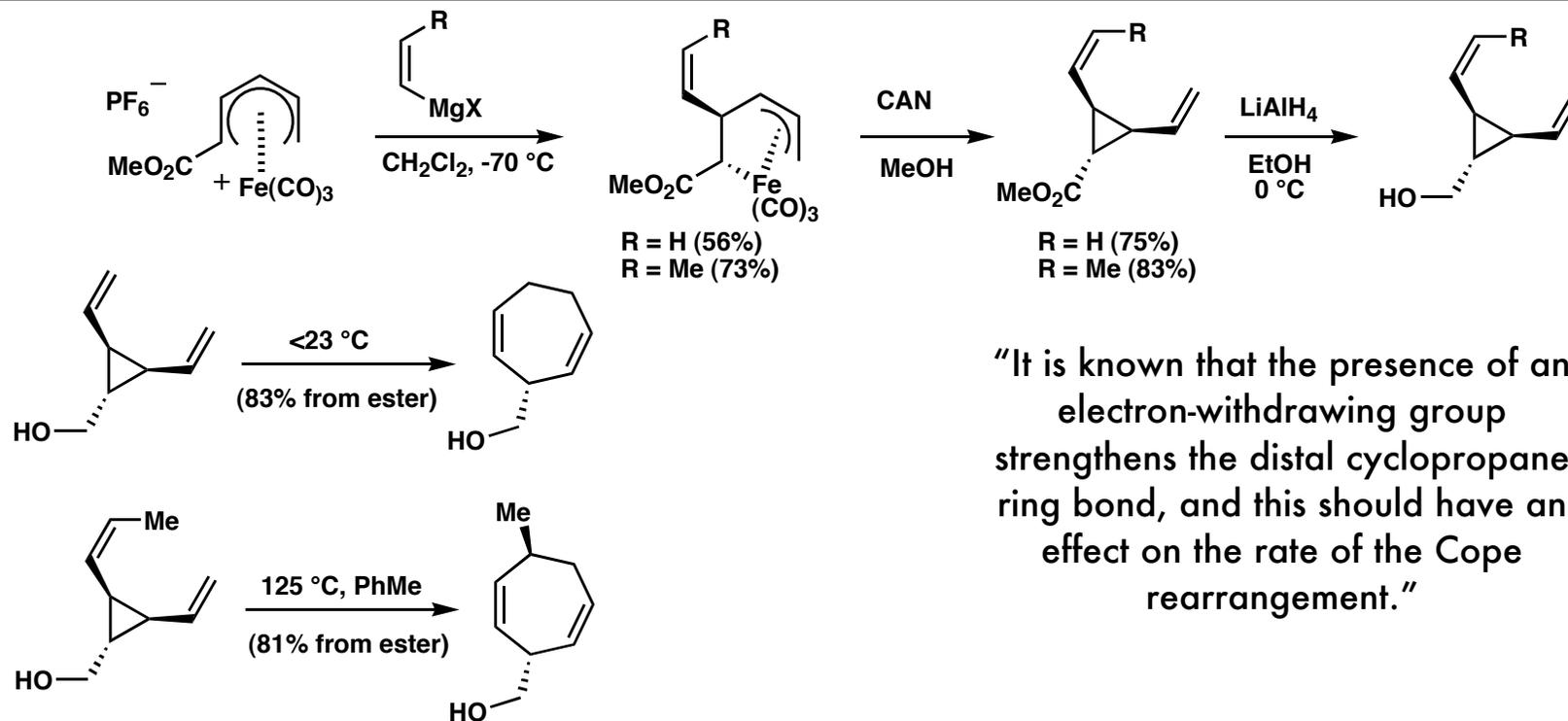
Malonate addition substrate scope



Yun, Y. K.; Godula, K.; Cao, Y.; Donaldson, W. A. *J. Org. Chem.* **2003**, *68*, 901-910

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Grignard additions to dienyliron cations



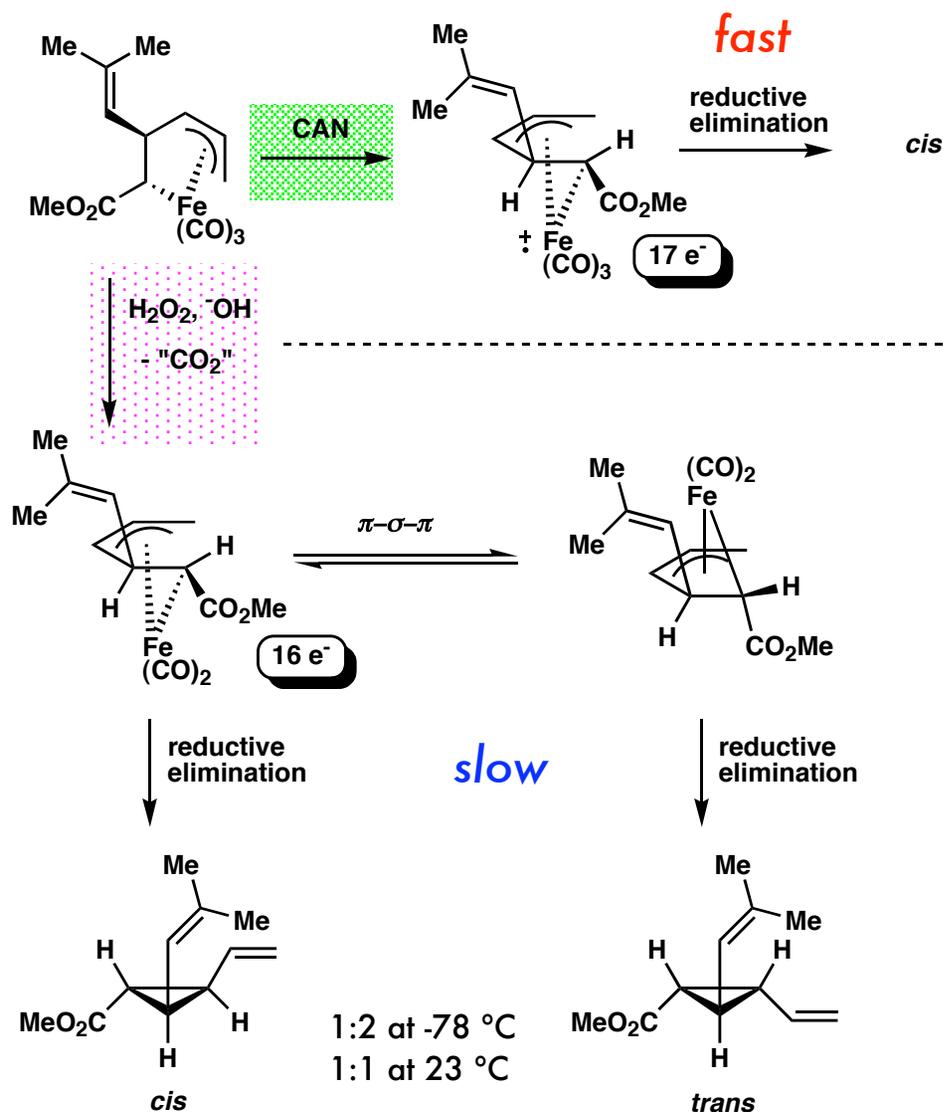
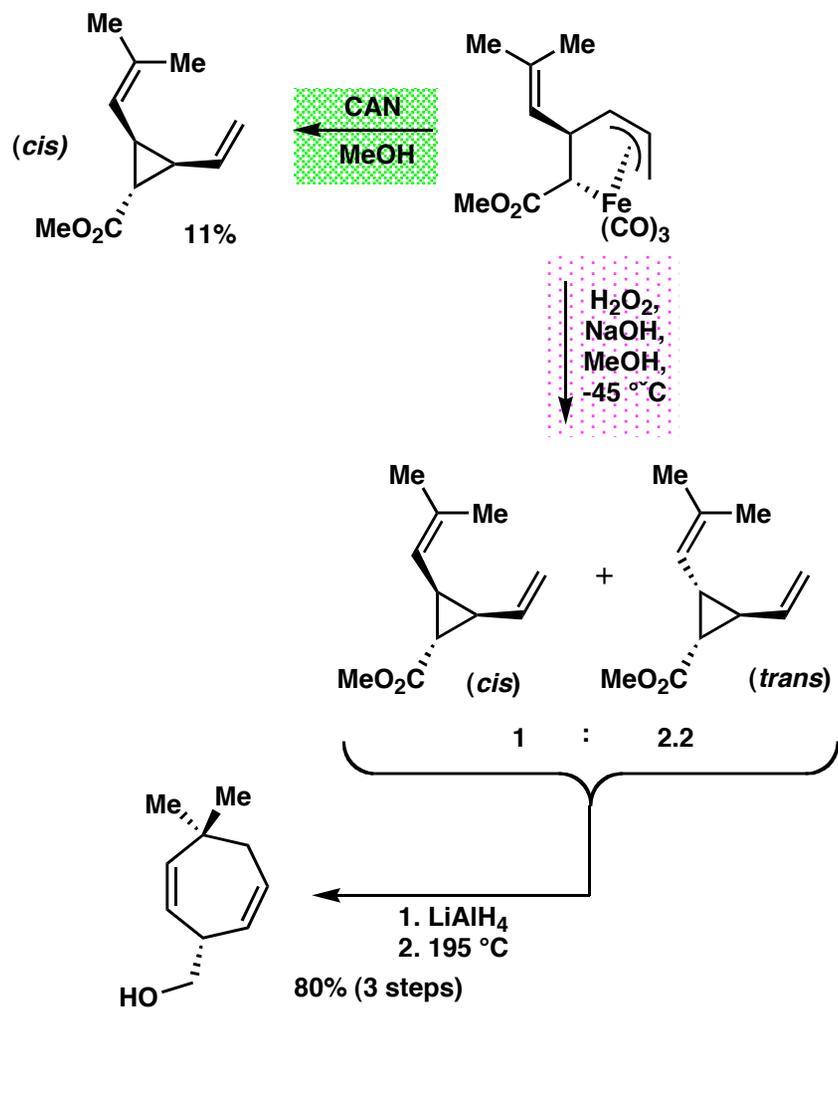
"It is known that the presence of an electron-withdrawing group strengthens the distal cyclopropane ring bond, and this should have an effect on the rate of the Cope rearrangement."

pentenediyl complex	oxidation conditions ^a	divinyl-cyclopropane	Cope conditions ^a	1,4-cycloheptadiene
 4c (42-52%)	A	 5c ^c	C	 6c (82%) ^b
 4d (38-49%)	A	not observed	C	 6d (33%) ^b
		...		

pentenediyl complex	oxidation conditions ^a	divinyl-cyclopropane	Cope conditions ^a	1,4-cycloheptadiene
 4e (71-76%)	A B	 5e (11%, <i>cis</i> only) 5e/e' (63-84%) ^d	D	 6e (80%) ^b
 4f (50%)	B	 5f (85%) ^e	E	 6f (76%) ^b

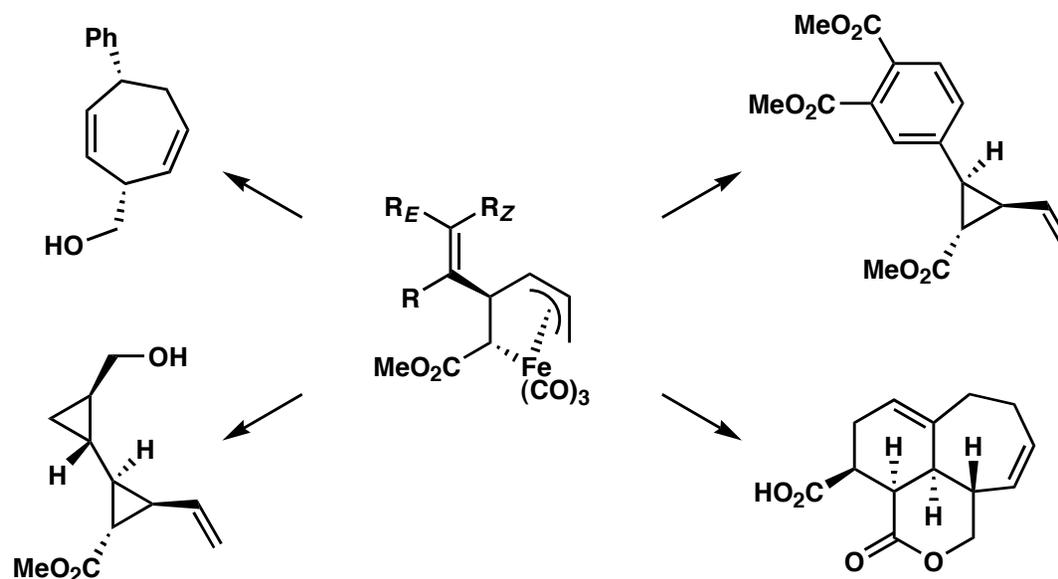
Wallock, N. J.; Donaldson, W. A. *Org. Lett.* **2005**, *10*, 2047-2049

Mechanistic Rationale:

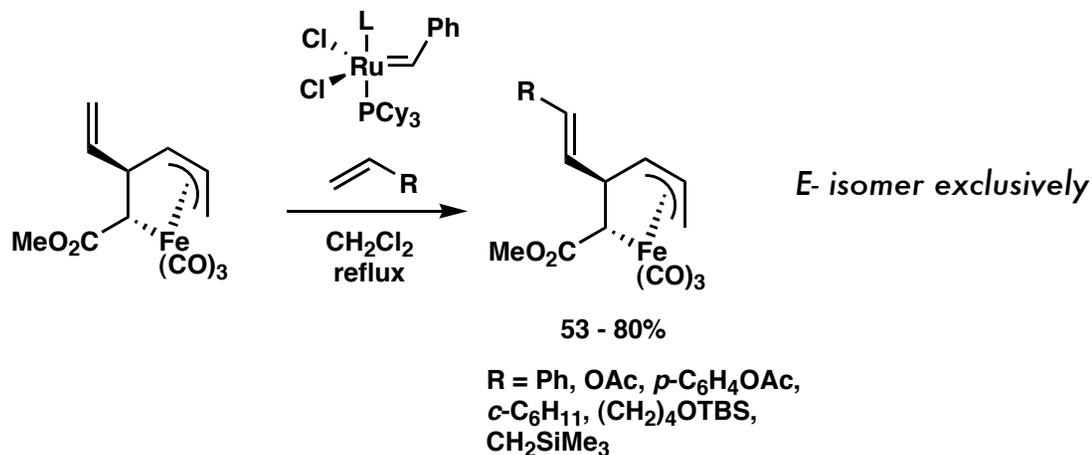


Wallock, N. J.; Bennett, D. W.; Siddiquee, T.; Haworth, D. T.; Donaldson, W. A. *Synthesis*, 2006, 21, 3639-3646.

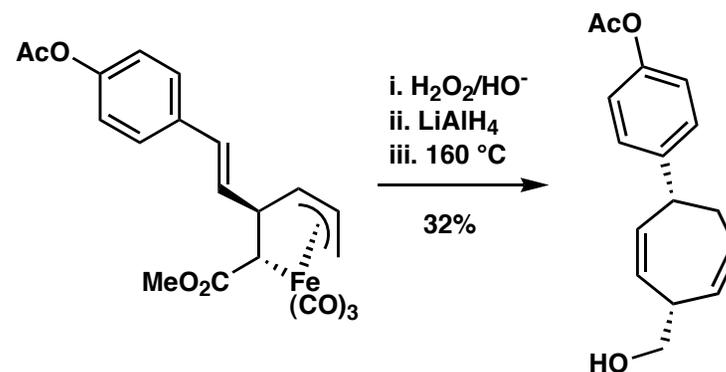
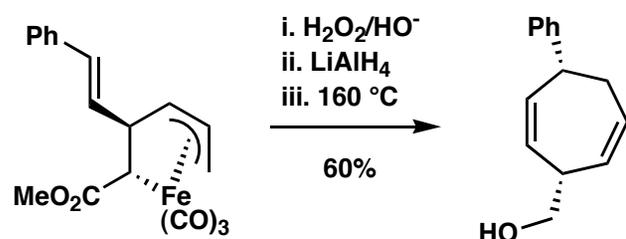
Current paper- scaffold diversification



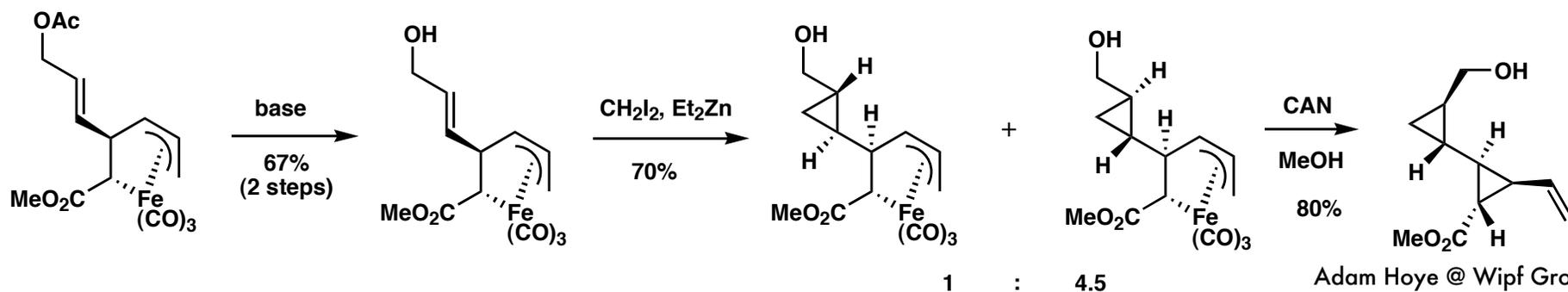
Metathesis strategy



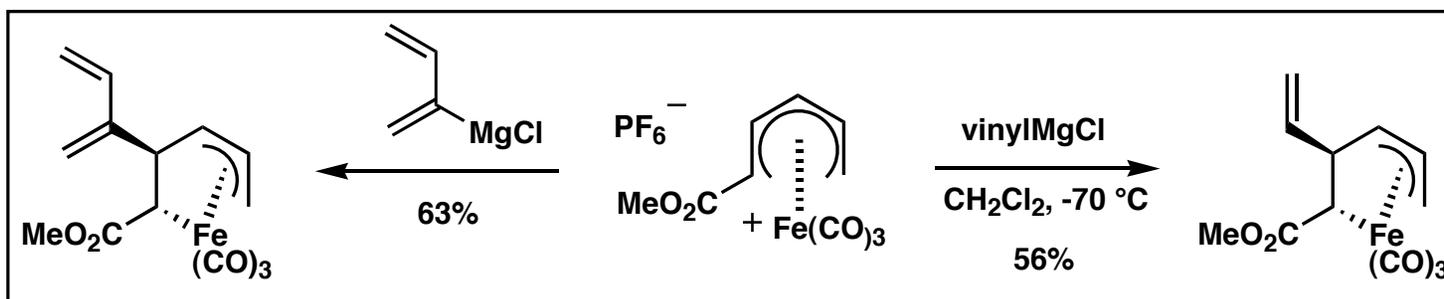
Divinylcyclopropane rearrangement:



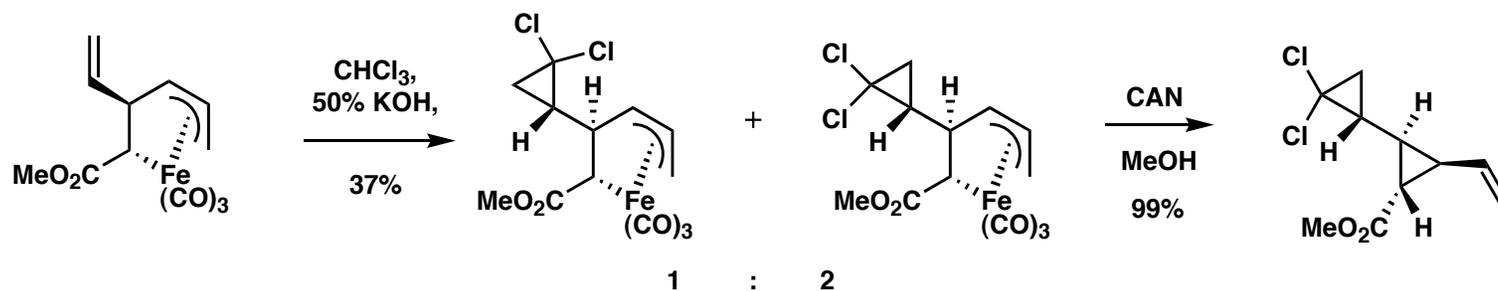
Cyclopropanation (Simmons-Smith):



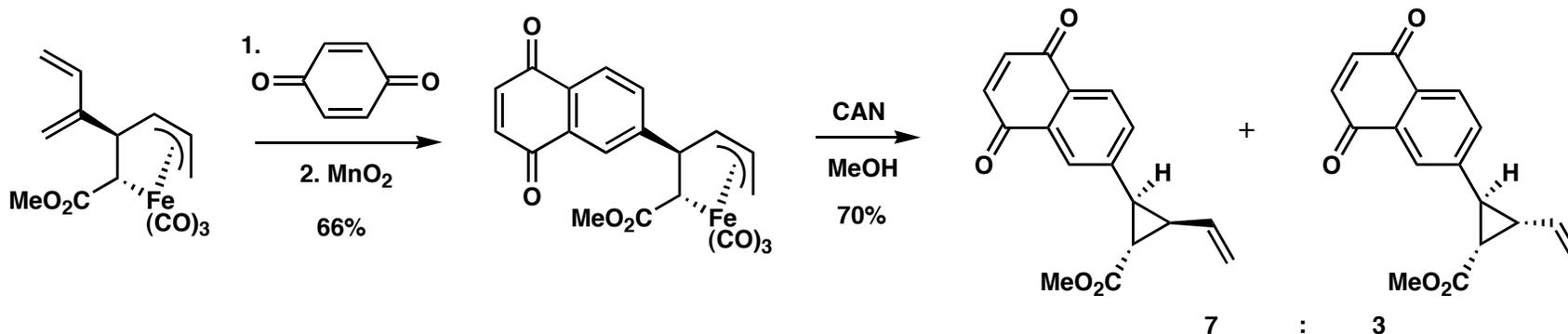
Grignard additions



Dichlorocyclopropanation:

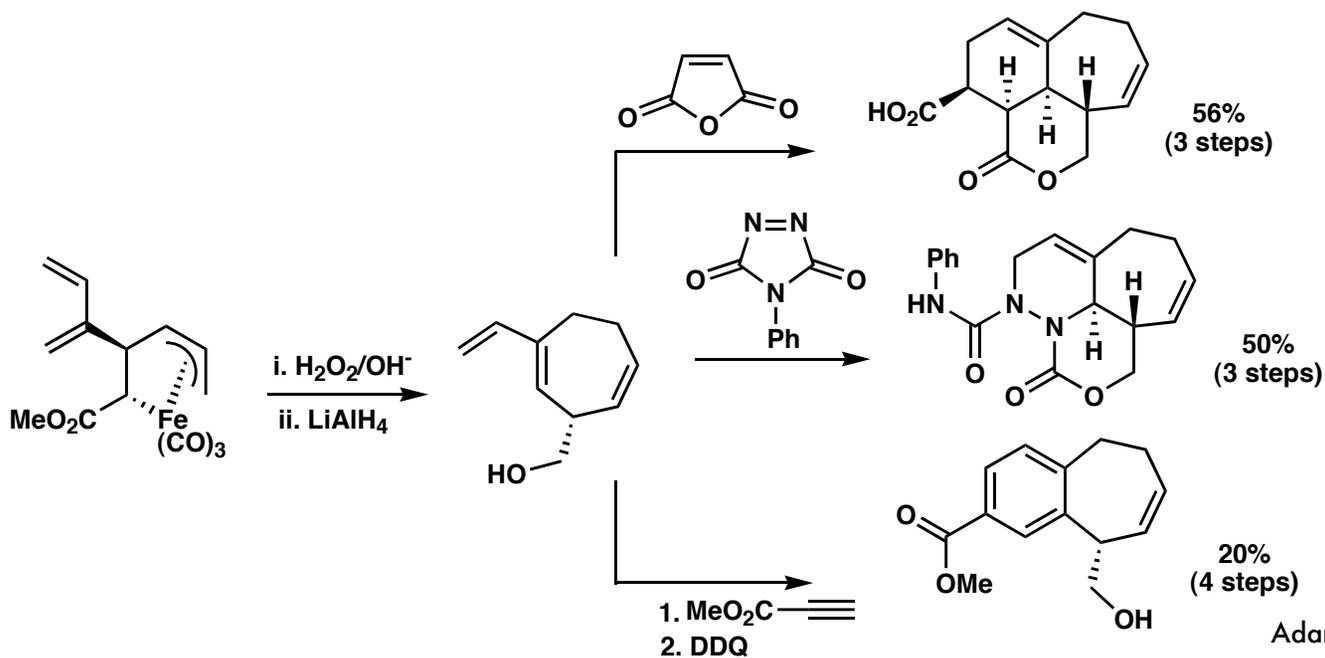
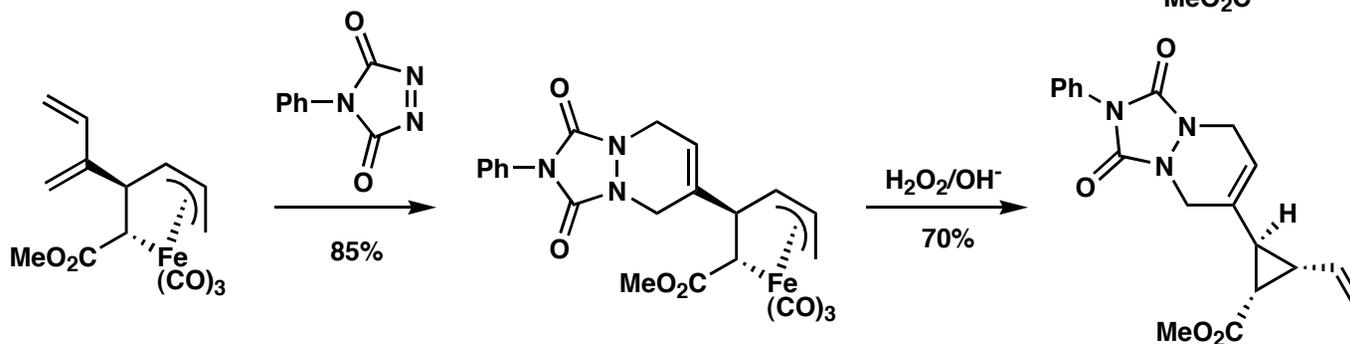
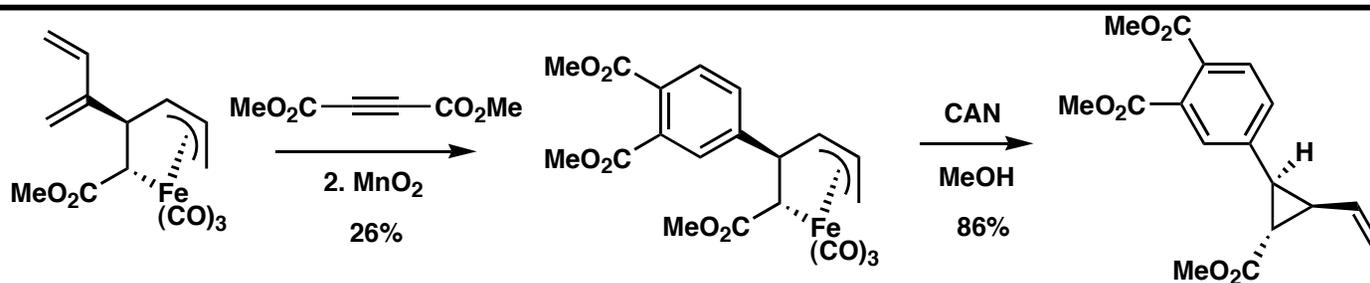


[4+2] Cycloadditions:



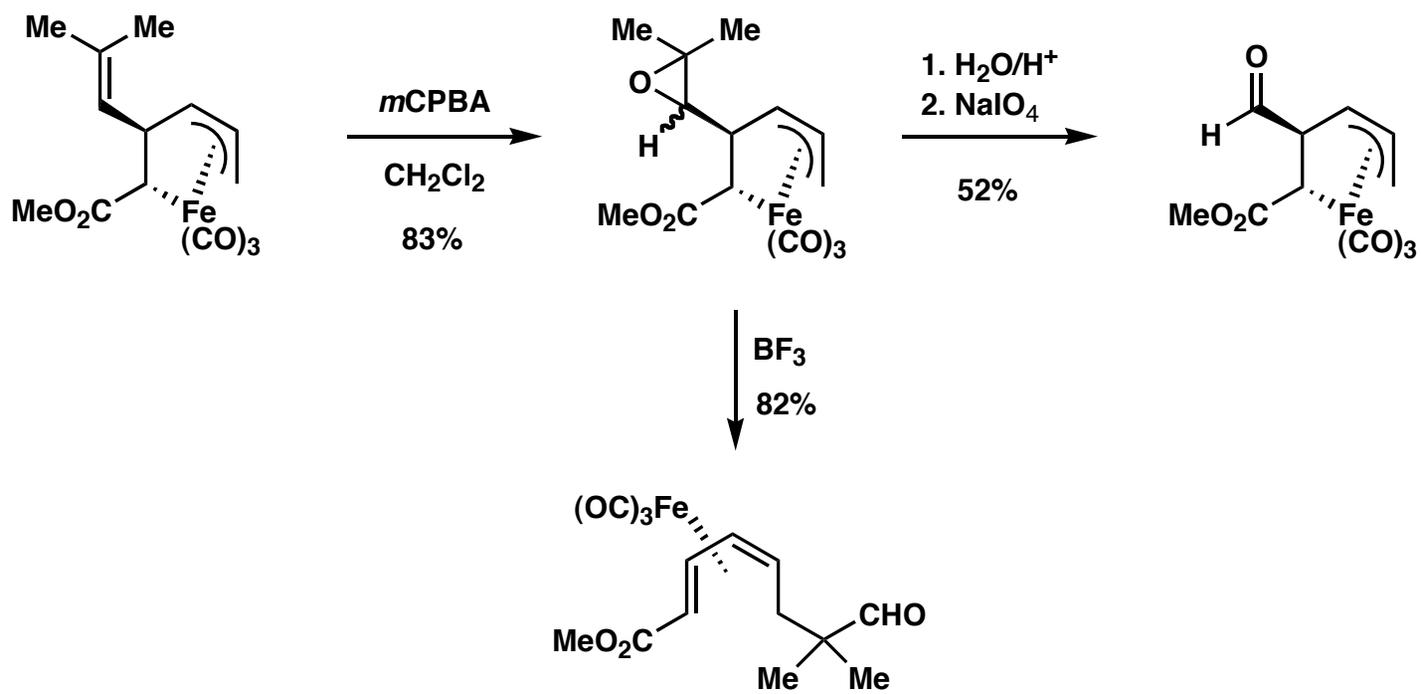
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Cycloadditions continued

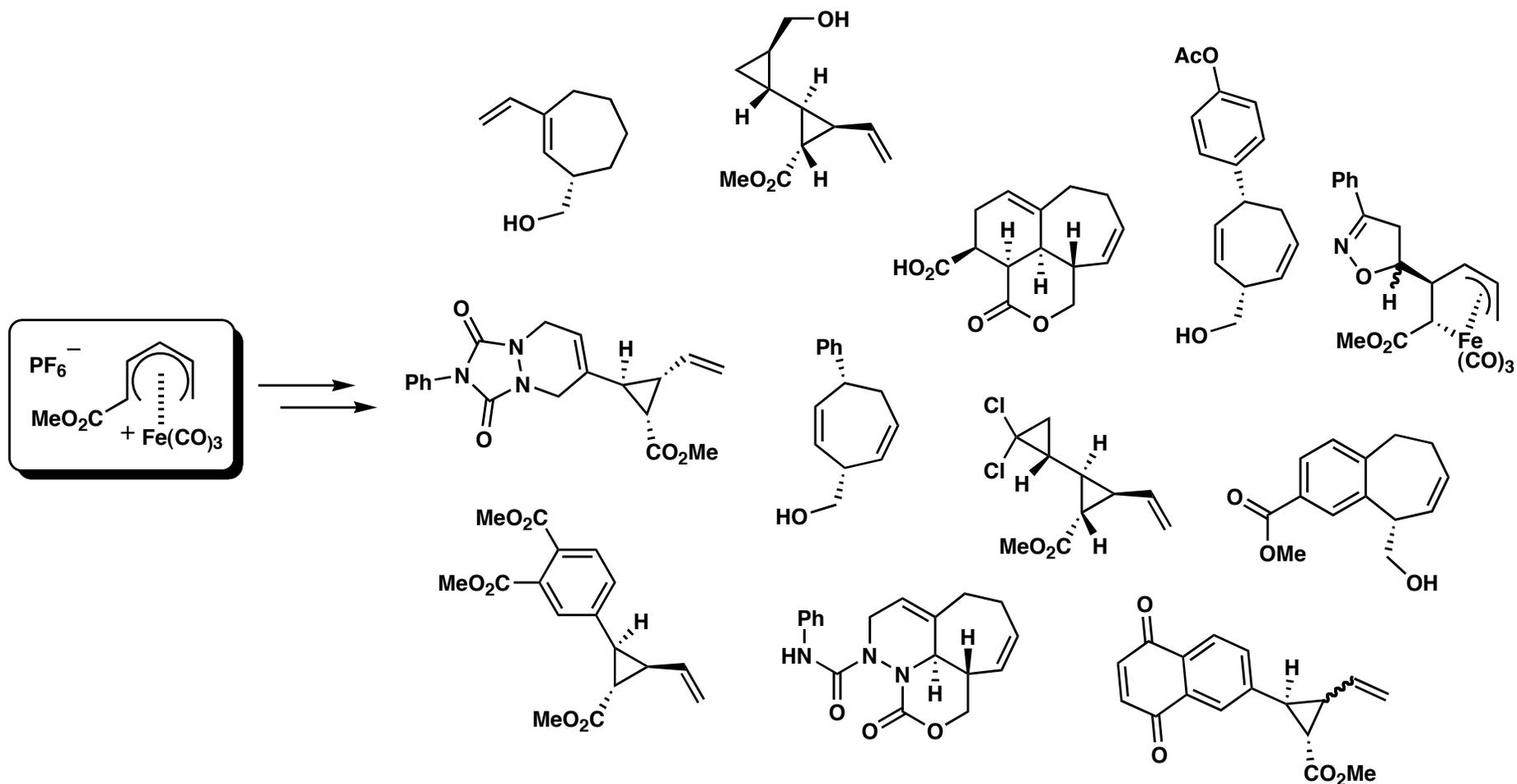


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Epoxidation strategy



Summary



-diverse molecular scaffolds accessible from a single organometallic intermediate
-tolerant of a variety of reaction conditions to access a broad range of
functionality with many points amenable to structural diversification